

# Detoxifying Oxidized Contaminants

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<http://environmentalbiotechnology.org>

# Acknowledgements

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- Workshop on Reducing Oxidized Contaminants, IWA Specialists Conference on Biofilms in Drinking Water, Arosa, Switzerland, August 2015.

# Detoxifying Oxidized Contaminants

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- ❖ **Contaminants:** harmful water pollutants.
- ❖ **Oxidized:** can receive electrons.
- ❖ The common characteristic of all oxidized contaminants is that they are bacterial **electron acceptors**. Opposite of biological instability, e.g., BOM or  $\text{NH}_4^+$ .
- ❖ Reduction yields a **harmless** solute or an easily removed solid → **detoxified**. No brine or residue that is a disposal challenge and cost.

# Examples of Oxidized Contaminants

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- ❖ **Nitrate** – agriculture and wastewater – reduced to  $N_2$
- ❖ **Perchlorate** – rocket fuel – reduced to  $Cl^-$  and  $H_2O$
- ❖ **Selenate** – coal-fired power plants, irrigation drainage, industry – reduced to solid  $Se^0$
- ❖ **Chromium(VI)** – mining areas and metals industries – reduced to  $Cr(OH)_3(s)$
- ❖ **Chlorinated solvents, like TCE** – everywhere – semi-conductors, metals working, dry cleaners, etc. – reduced to ethene and  $Cl^-$
- ❖ **Bromate** – from ozone use in water treatment, bromine industry – reduced to  $Br^-$
- ❖ **Radionuclides, including Pu, U, and Np** – Nuclear-weapons sites, mines, and mills – reduced to, e.g.,  $UO_2(s)$

# What are the necessary conditions?

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- Anoxia, or dissolved oxygen removed
- Excellent biomass retention as a biofilm
- Satisfactory pH, usually near neutral
- Availability of nutrients, particularly P
- **Supply sufficient electron donor**
  - Organic compound (for heterotrophs)
  - Hydrogen gas ( $H_2$ ) or S (for autotrophs)

# Heterotrophic vs Autotrophic

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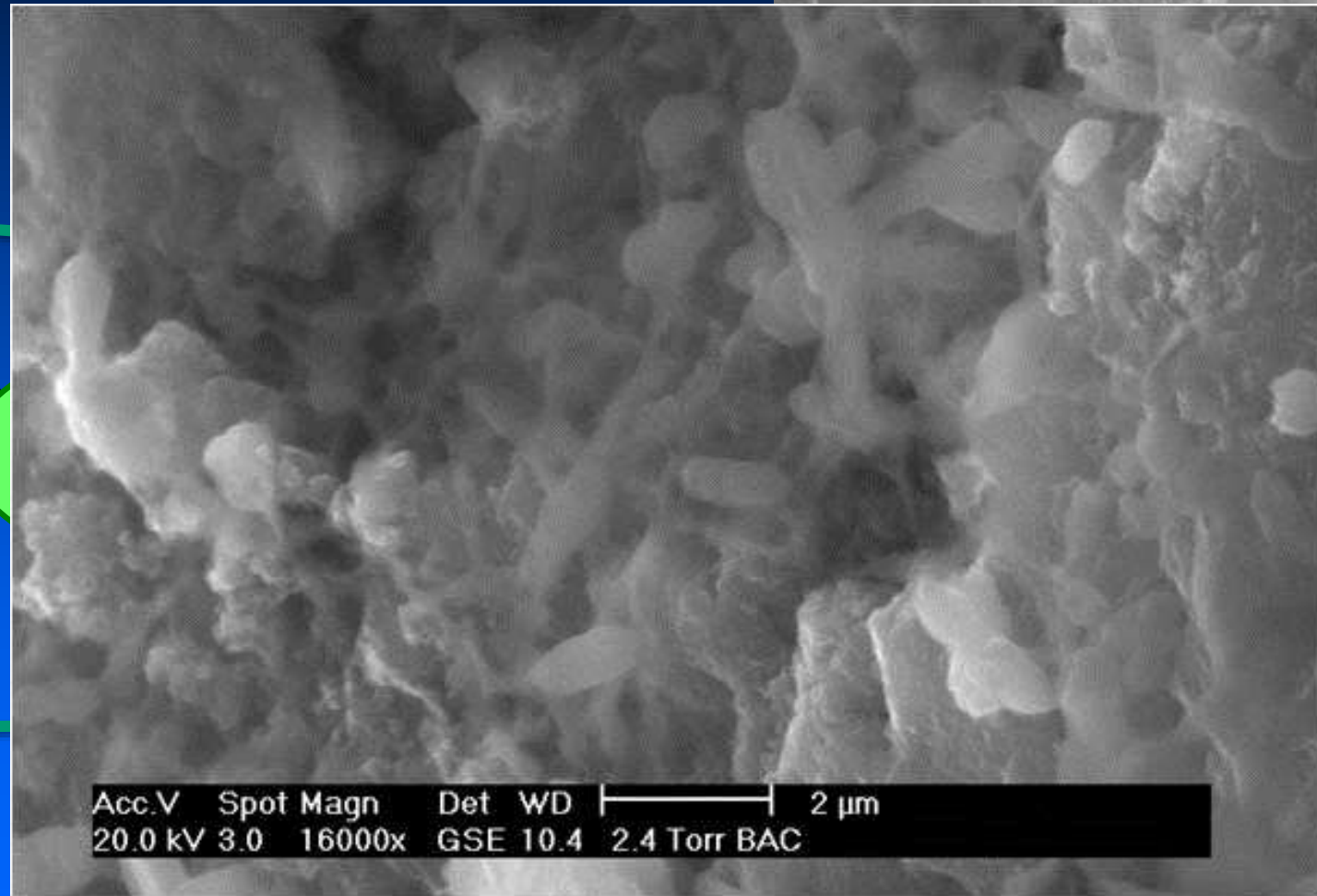
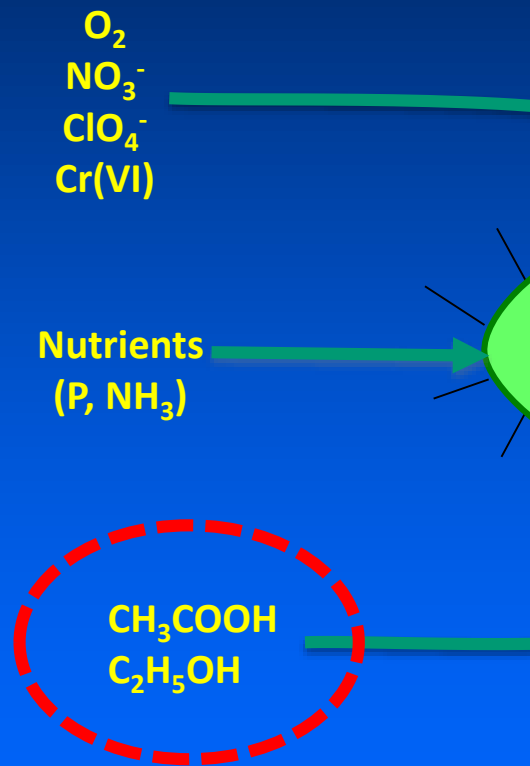
- **Heterotrophic** processes use an ORGANIC C source
- **Autotrophic** processes use an INORGANIC C source (e.g.,  $\text{CO}_2$  or  $\text{HCO}_3^-$ )
- Typically,
  - a microorganism that uses an organic donor also uses it as the C source
  - A microorganism that uses an inorganic donor also uses an inorganic C source

# What should you know?

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- What is the status of heterotrophic and autotrophic biofilm process available for reducing oxidized contaminants?
- What are the pros and cons of each approach?

# Heterotrophic Processes



20.0 kV 4.0 140x GSE 10.8 0.6 Torr BAC

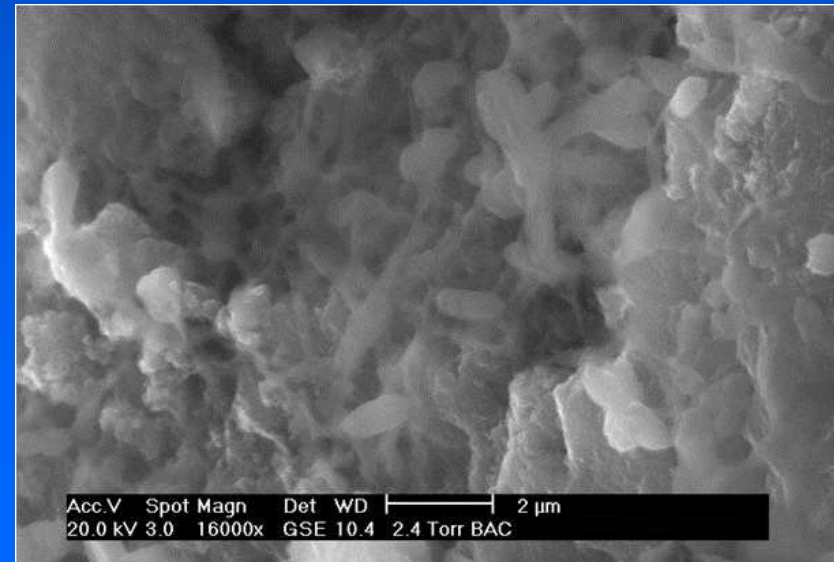
Media: GAC, anthracite, sand, expanded clay



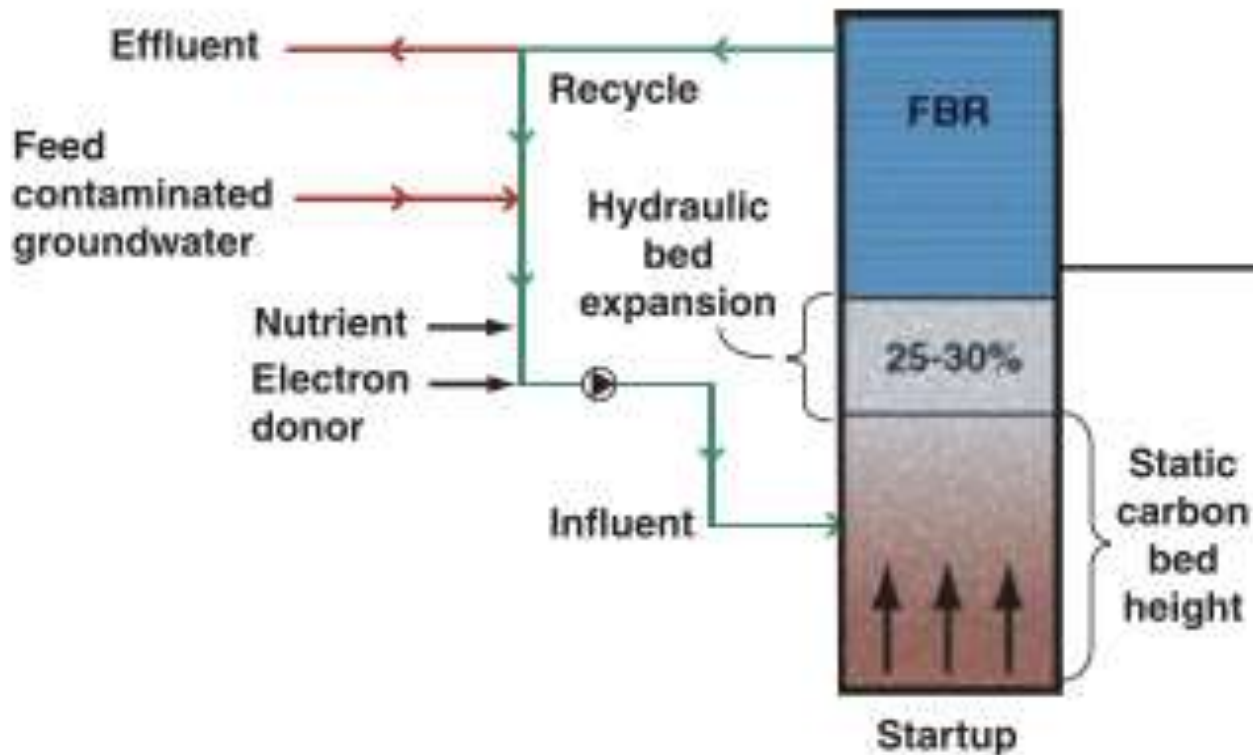
# General organic carbon considerations

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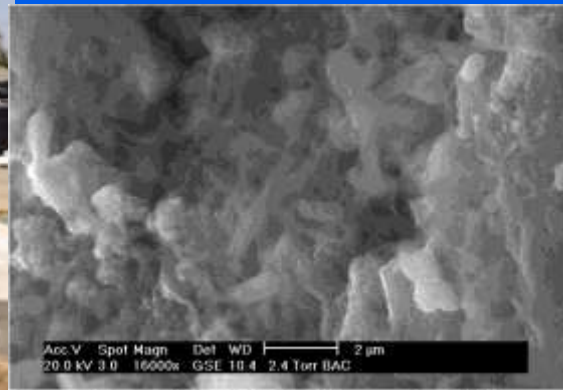
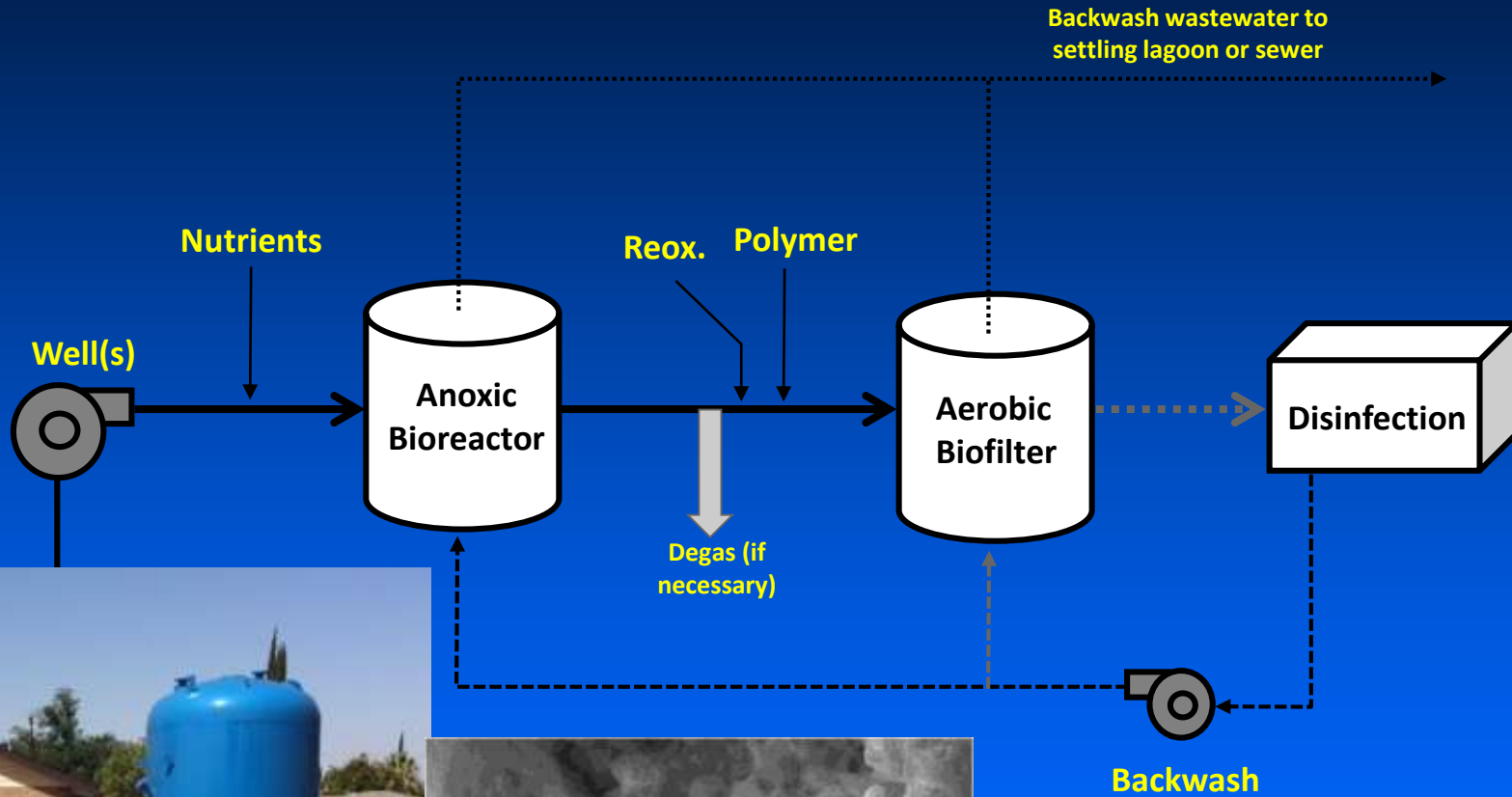
- High biomass growth / fast kinetics
- Expensive/limited certified suppliers
- Disinfection by-product precursor
- May need specialized storage



# Envirogen fluidized-bed bioreactor



# biotta™ fixed bed bioreactor Two-Stage Fixed Bed



# Autotrophic Processes

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- Autotrophic processes use an INORGANIC C source
- Typically, *inorganic donors* use *inorganic C sources*

# Examples of Inorganic Donors

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- Hydrogen gas ( $\text{H}_2$ )
- Reduced sulfur ( $\text{S}^0$ ,  $\text{H}_2\text{S}$ ,  $\text{S}_2\text{O}_3^{2-}$ )
- Reduced iron ( $\text{Fe}^{2+}$ )

# Advantages and Disadvantages of Autotrophy

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- Lower biomass yields, but slower growth rates
- Often sparsely soluble – no residual, but need special delivery strategy
- Typically have no or low toxicity, but may have special handling needs

# H<sub>2</sub> as an autotrophic donor

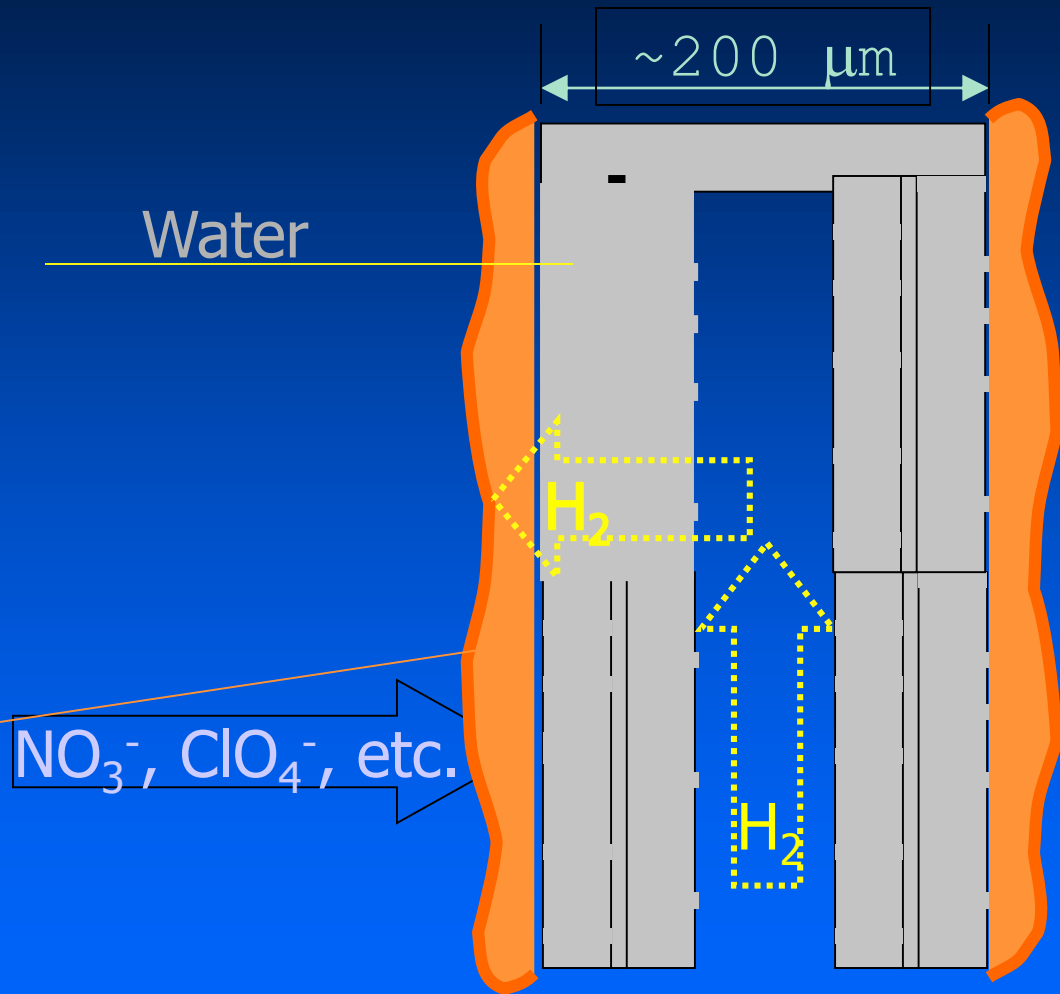
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- Low cost
- High electron density per unit mass
- Can be produced on site
- Can reduce the wide range of oxidized contaminants (“universal electron donor”)

# The Membrane Biofilm Reactor (MBfR) for delivering H<sub>2</sub> to the biofilm

POLYPROPYLENE  
MEMBRANE!

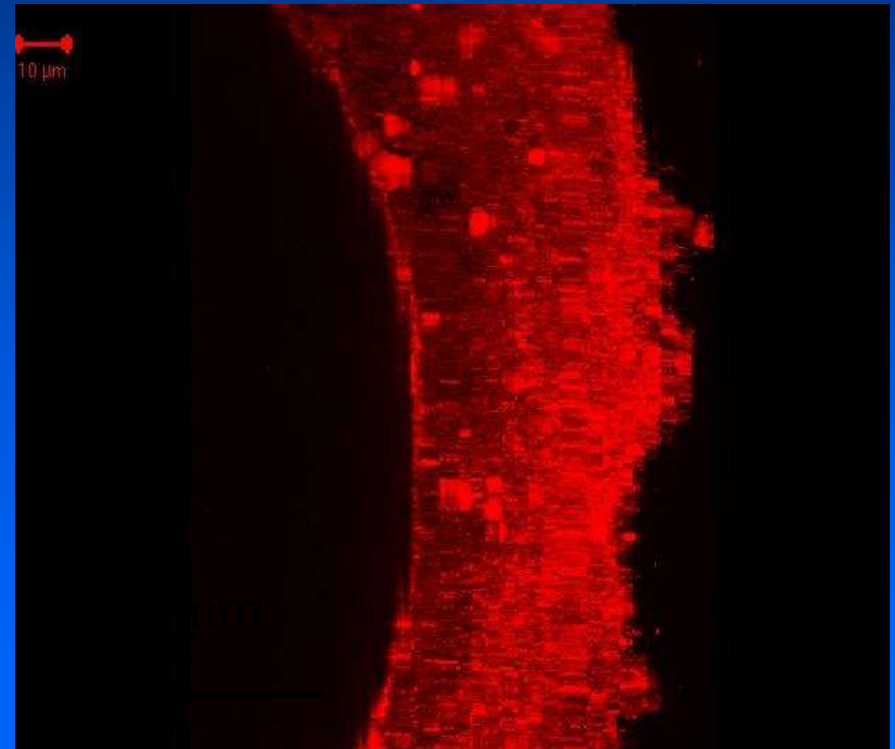
Biofilm “reactive barrier”  
“Supply on demand”  
H<sub>2</sub>-use efficiency  
nearly 100%



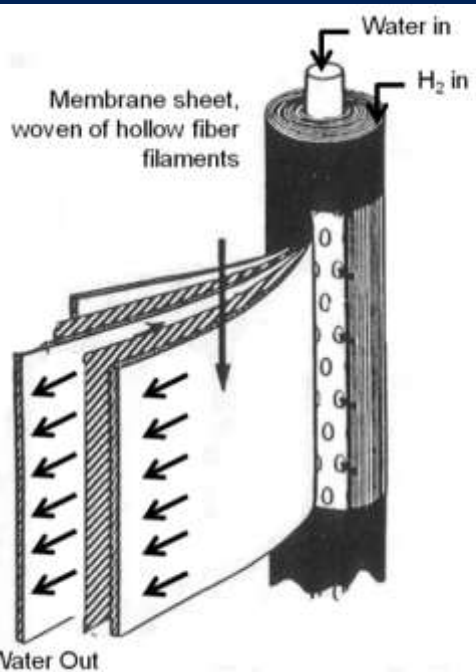


# What do the hollow-fibers and biofilms look like?

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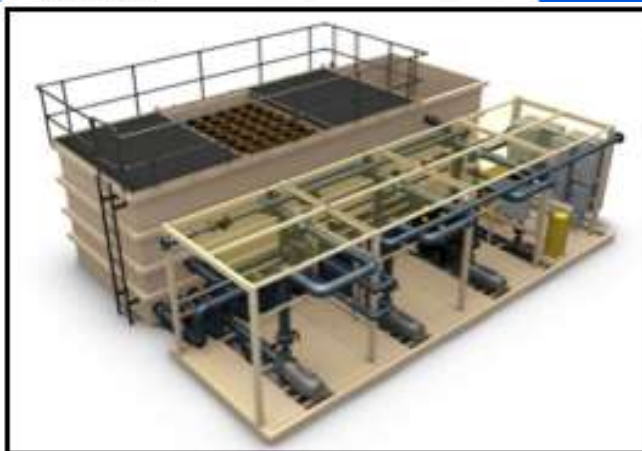
# Pilot- and Commercial-scale MBfR – ARoNITE by APTwater



Spiral wound sheets of polypropylene fibers



Inside the tank of modules for one stage of a pilot



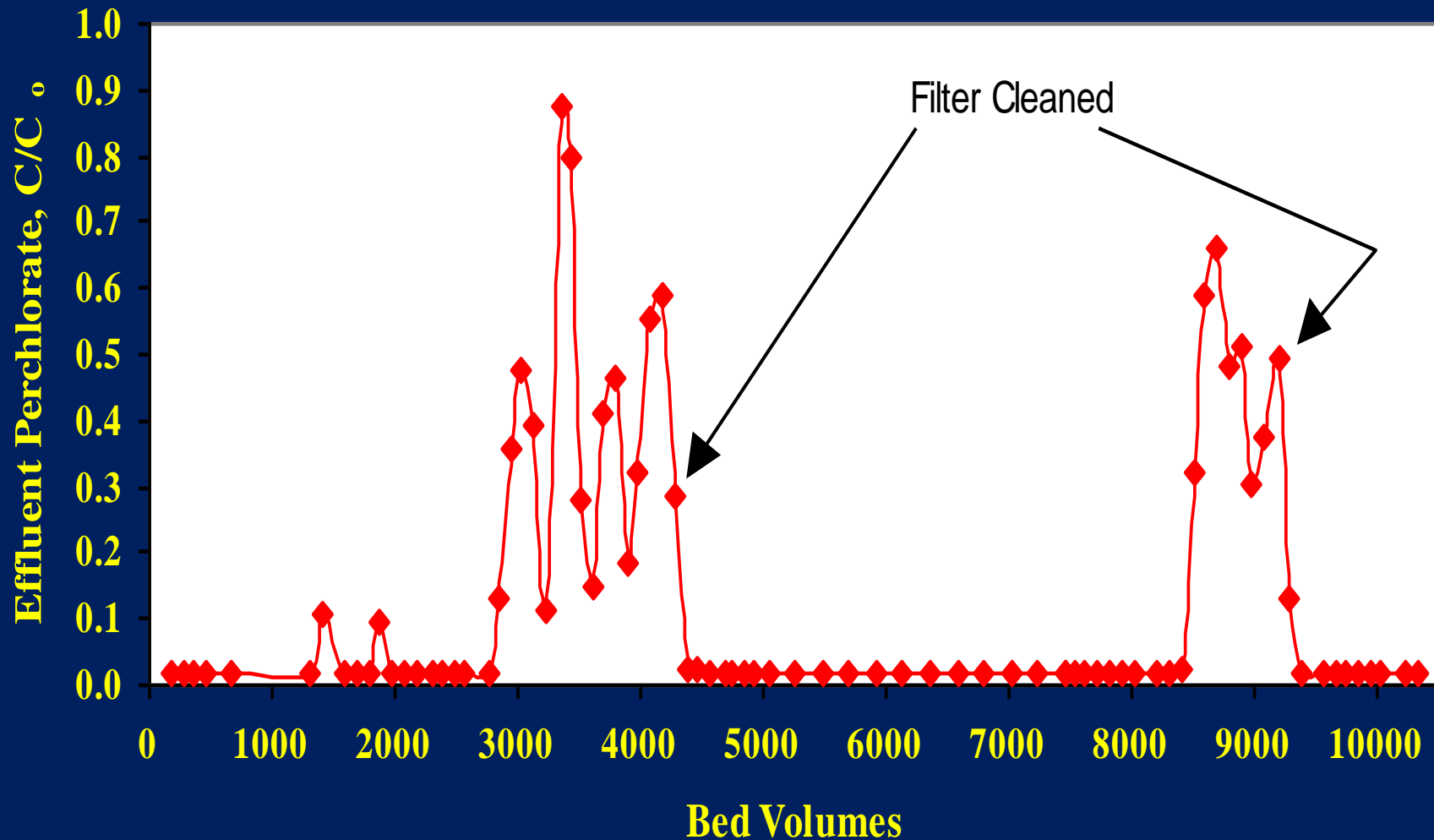
Full-scale multi-stage  
ARoNITE in Calliforina

# Take-Home Lessons and Pressing Issues

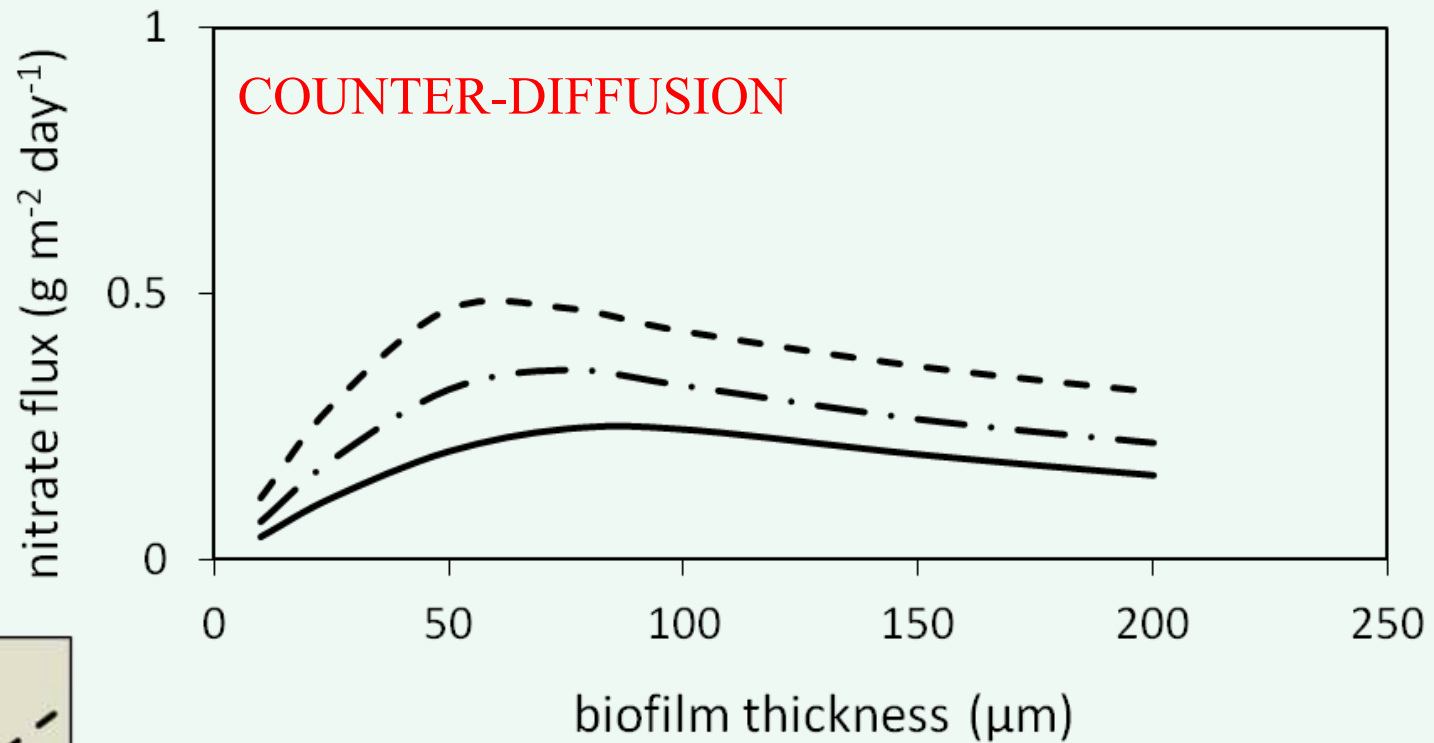
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- 1) Biomass management is critical
  - a) To prevent excessive biofilm accumulation and head loss
  - b) “Backwashing” methods are well developed

# Heterotroph Fixed-Bed Example



# Can have too much autotrophic biofilm



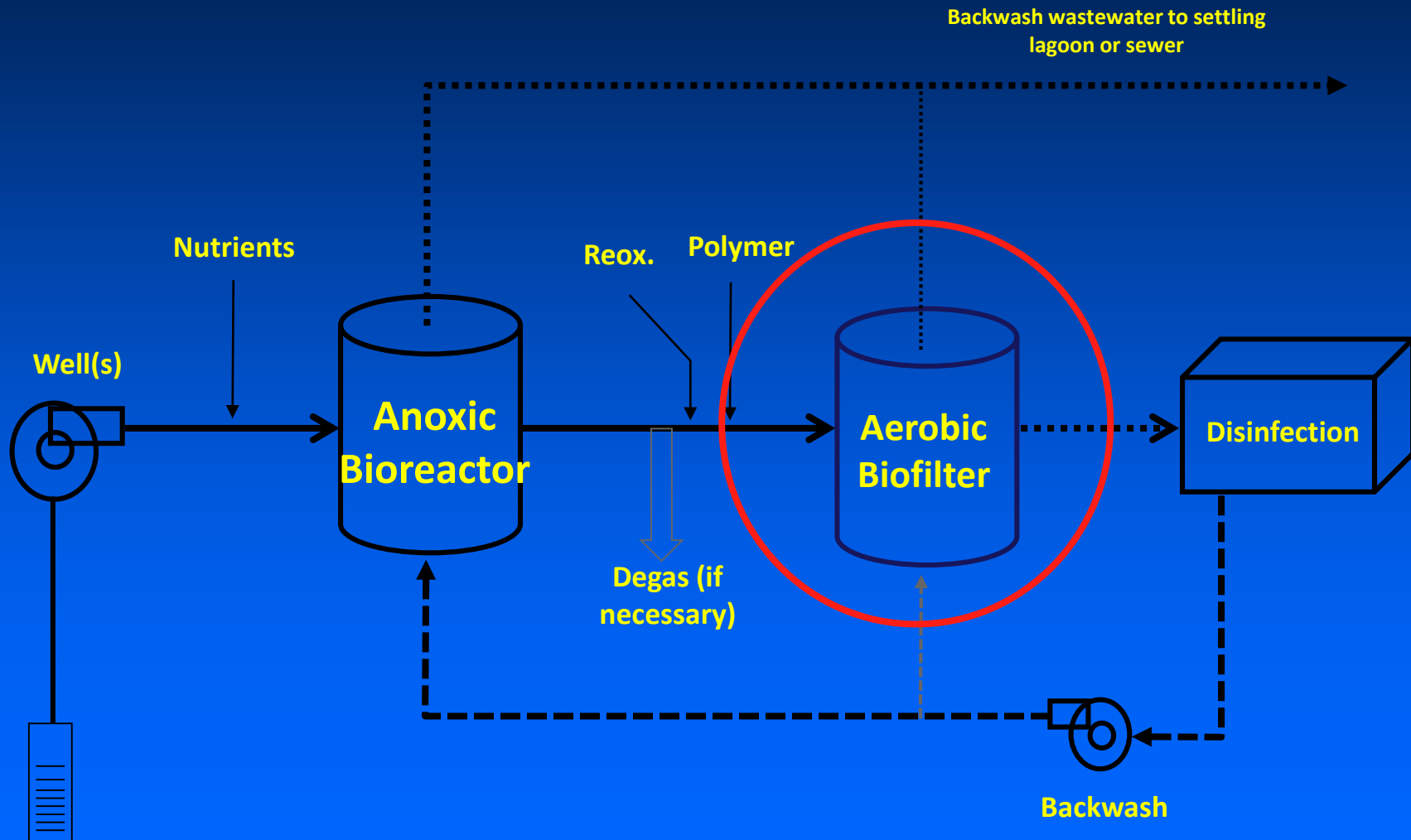
Modeling results from Rob Nerenberg. Lack of donor and acceptor overlap with a too-thick biofilm

# Take-Home Lessons and Pressing Issues

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- 1) Biomass management is critical
  - a) To prevent excessive biofilm accumulation and head loss
  - b) “Backwashing” methods are well developed
- 2) Accurate delivery of the donor is essential
  - a) Avoid over- or under-dosing for organic donors
  - b) Ensure enough delivery capacity for H<sub>2</sub>

# Aerobic Post-treatment with Heterotrophic Processes



# H<sub>2</sub>-Based On-demand Delivery

POLYPROPYLENE  
MEMBRANE!

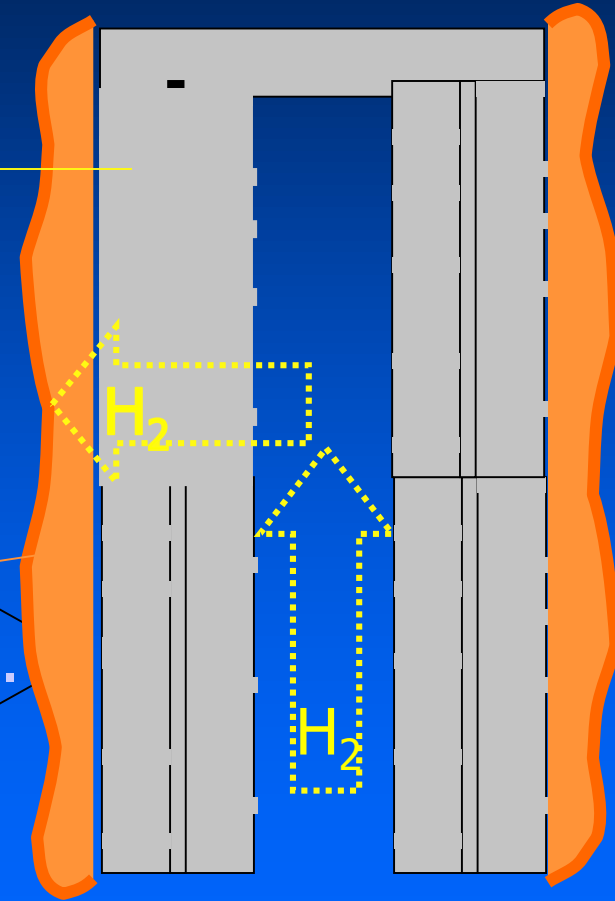
Water

Biofilm “reactive  
barrier”

NO<sub>3</sub><sup>-</sup>, ClO<sub>4</sub><sup>-</sup>, etc.

“Supply on demand”

H<sub>2</sub>-use efficiency  
nearly 100%





# Take-Home Lessons and Pressing Issues

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- 1) Biomass management is critical
  - a) To prevent excessive biofilm accumulation and head loss
  - b) “Backwashing” methods are well developed
- 2) Accurate delivery of the donor is essential
  - a) Avoid over- or under-dosing for organic donors
  - b) Ensure enough delivery capacity for H<sub>2</sub>
- 3) pH control needed due to base release.
- 4) Automated operation and on-line monitoring are needed and being implemented
- 5) Removing multiple oxidized contaminants is a major advantage and challenge

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